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General Questions

Practice Questions

Q1. What trend is observed in atomic radius across a period in the periodic table?

A. Increases
B. Decreases
C. Remains constant
D. Varies randomly

Solution: Atomic radius decreases across a period due to increasing nuclear charge attracting electrons closer.

Q2. Which of the following elements has the largest atomic radius?

A. Lithium
B. Sodium
C. Potassium
D. Rubidium

Solution: Rubidium has the largest atomic radius due to its position in Group 1 and being the lowest in the group.

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