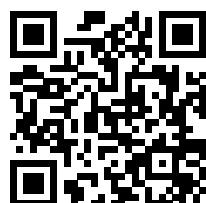




SoulShift - Educational Q&A Platform

General Questions

Practice Questions



Q1. What trend is observed in atomic radius across a period in the periodic table?

- A. Increases
- B. Decreases
- C. Remains constant
- D. Varies randomly

Solution: Atomic radius decreases across a period due to increasing nuclear charge attracting electrons closer.

Q2. Which of the following elements has the largest atomic radius?

- A. Lithium
- B. Sodium
- C. Potassium
- D. Rubidium

Solution: Rubidium has the largest atomic radius due to its position in Group 1 and being the lowest in the group.



